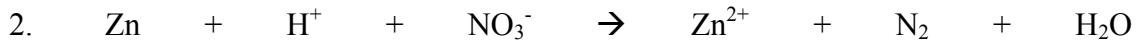


Academic Chemistry Oxidation Number Method Wkst Name: _____

Balance the following redox reaction using the oxidation number method. Identify the oxidizing agent (OA), the reducing agent (RA), the element oxidized, and the element reduced.



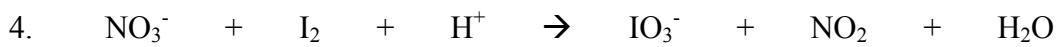
Element Oxidized _____ Element Reduced _____ RA _____ OA _____



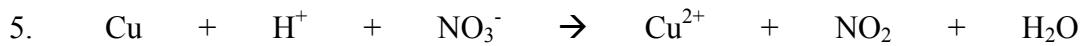
Element Oxidized _____ Element Reduced _____ RA _____ OA _____



Element Oxidized _____ Element Reduced _____ RA _____ OA _____



Element Oxidized _____ Element Reduced _____ RA _____ OA _____



Element Oxidized _____ Element Reduced _____ RA _____ OA _____



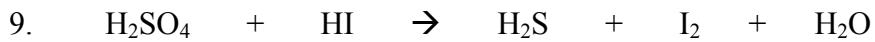
Element Oxidized _____ Element Reduced _____ RA _____ OA _____



Element Oxidized _____ Element Reduced _____ RA _____ OA _____



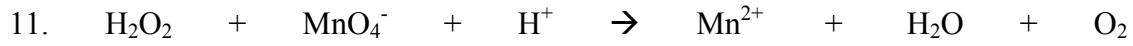
Element Oxidized _____ Element Reduced _____ RA _____ OA _____



Element Oxidized _____ Element Reduced _____ RA _____ OA _____



Element Oxidized _____ Element Reduced _____ RA _____ OA _____



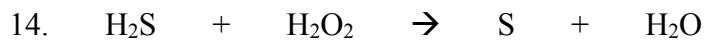
Element Oxidized _____ Element Reduced _____ RA _____ OA _____



Element Oxidized _____ Element Reduced _____ RA _____ OA _____



Element Oxidized _____ Element Reduced _____ RA _____ OA _____



Element Oxidized _____ Element Reduced _____ RA _____ OA _____



Element Oxidized _____ Element Reduced _____ RA _____ OA _____